(ECC01-NE-01E)

- One mole of C2H4 reacts with three moles of O2. 56g of C2H4 will react with
 - A) 6 moles of O2
 - B) 3.61 x 1024 molecules of O2
 - C) 134.4 dm3 of O2
 - D) All of these

(ECC01-NE-02E)

- 5894 is written to base 10 as:
 - A) 5.894 x 10-3
 - B) 5.894 x 104
 - C) 5.894 x 102
 - D) 5.894 x 103

(ECC01-NE-03E)

- 870.0 have _____ significant figures.
 - A) Two
 - B) Three
 - C) Four
 - D) Six

(ECC01-NE-04E)

- A chemical formula based on actual number of molecule is called ______.
 - A) Structural
 - B) Molecular
 - C) Empirical
 - D) None





(ECC01-NE-05E)

- Which of the following is a single pure compound?
 - A) Air
 - B) Steam
 - C) Brass
 - D) Sea Water

(ECC01-NE-06E)

- Formula of ferric sulphate is:
 - A) FeSO4
 - B) Fe(SO4)3
 - C) Fe2(SO4)3
 - D) Fe2SO4

(ECC01-NE-07E)

- A mixture whose composition is uniform throughout the mixture is called:
 - A) Heterogeneous
 - B) Homogenous
 - C) Homonuclear
 - D) Heteronuclear

(ECC01-NE-08E)

- The number of moles present in 6gms of carbon is:
 - A) 2
 - B) 0.5
 - C) 5
 - D) 1





(ECC01-NE-09E)

- 1 cm3 is equal to _____ m3:
 - A) 10⁻²
 - B) 10⁻⁶
 - C) 10^{6}
 - D) 10⁴

(ECC01-NE-10E)

- ____16 gm of O2 contains
 - A) 3.01 X 10²³ molecules
 - B) 6.02 X 10²³ atoms
 - C) 6.02 X 10²⁴ molecules
 - D) 3.01 X 10⁻²³ atoms

(ECC01-NE-11E)

- The -ve charged particles is called:
 - A) Anion
 - B) Cation
 - C) Radical
 - D) Atom



(ECC01-NE-12E)

- 20g of a monoatomic gas occupies 44.8 dm3 at STP. Its atomic weight is
 - A) 5g
 - B) 10g
 - C) 40g
 - D) 20g





(EPC01-NE-13E)

- a pure chemical compound always contains the same elements combined in the same ratio by weight is:
 - A) Law of definite proportions
 - B) Law of multiple proportions
 - C) Law of mass action
 - D) Law of equilibrium

(ECC01-NE-14E)

- Any charged particle is called:
 - A) Atom
 - B) Molecule
 - C) Ion
 - D) Mixture

(ECC01-NE-15E)

- The no. of moles present in 1200 cm³ of a gas at S.T.P are:
 - A) 0.52 moles
 - B) 1.51 moles.
 - C) 0.053 moles
 - D) 0.053 moles









Answer key	
1	D
2	D
3	С
4	В
5	В
6	С
7	В
8	В
9	В
10	A
11	Α
12	В
13	Α
14	С
PA RHO	AUR C



